1. COMMUNICATING CHEMICAL REACTIONS

UNIT 2 CHEMICAL REACTIONS

CH30S MR. WIEBE

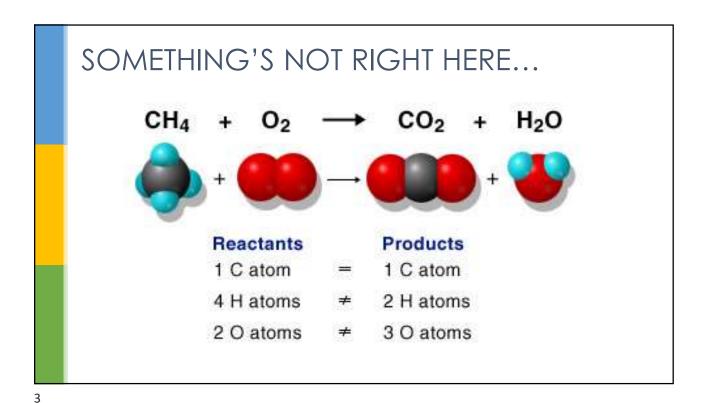
1

COMMUNICATING REACTIONS

Methane gas (carbon tetrahydride) combusts with oxygen gas from the air to produce carbon dioxide and water vapour.

WORD EQUATION:

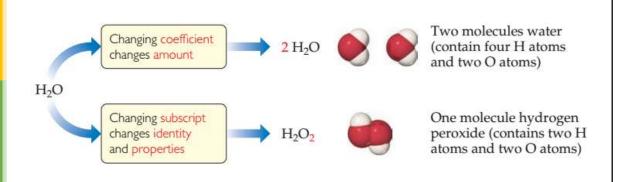
FORMULA EQUATION:

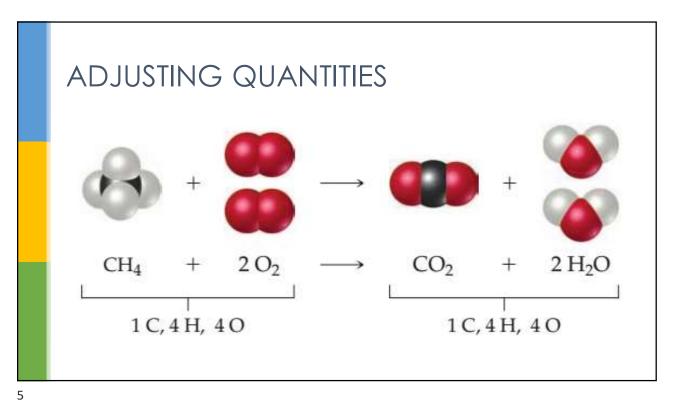


BALANCING A FORMULA EQUATION

• For a formula equation to be correct, it must be BALANCED.

 MULTIPLIERS are added in front of each formula. These multipliers are called <u>COEFFICIENTS</u>.





GENERAL BALANCING PRINCIPLES

$$AI + O_2 \rightarrow AI_2O_3$$

$$_$$
 Na(OH) + $_$ Fe(NO₃)₃ \rightarrow $_$ Na(NO₃) + $_$ Fe(OH)₃

$$\underline{\hspace{0.5cm}}$$
 $C_2H_6 + \underline{\hspace{0.5cm}}$ $O_2 \rightarrow \underline{\hspace{0.5cm}}$ $CO_2 + \underline{\hspace{0.5cm}}$ H_2O

	FX	Α	NΛ	P	F
- 1		/ \	1 V I	1 II I	

A strip of aluminum reacts with copper(II) chloride to produce copper and aluminum chloride.

Words		
Formulas		
Pictures		
Balanced Equation		